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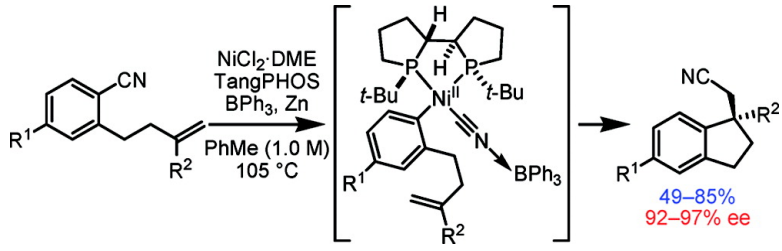

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Asymmetric Intramolecular Arylcyanation of Unactivated Olefins via C−CN Bond Activation

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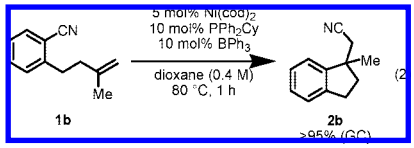
The development of transition metal-catalyzed transformations involving C−C bond activation is an emerging area in organic synthesis.[1] Methods where C−C bond activation is coupled with alkene insertion hold the potential to establish two new C−C bonds and up to two new stereogenic centers in a single operation. In this context, Nakao and Hiyama have disclosed that aryl nitriles can be activated and the resulting fragments can be added across alkynes[2] and strained alkenes such as norbonene[3] using nickel(0) catalysis.[4] Inspired by these important studies, we have explored the arylcyanation of unactivated olefins via C−CN bond activation. We disclose here the identification of catalytic asymmetric intramolecular olefin arylcyanations, providing indanes with quaternary carbon stereogenic centers from readily available benzonitrile precursors in good yields and high enantioselectivities.[5,6]

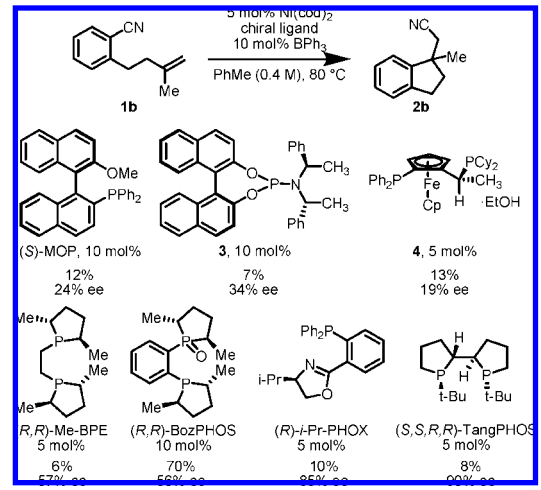
Initial studies focused on identifying reaction conditions for intramolecular olefin arylcyanation in a racemic manifold. Treatment of monosubstituted olefin **1a** to the conditions reported for alkyne arylcyanation (eq 1)[2] led only to partial olefin isomerization. In contrast, 1,1-disubstituted olefin **1b** underwent arylcyanation under the same conditions to afford indane **2b** in 21% yield. Systematic investigation led to identification of solvent, phosphine ligand, and particularly added Lewis acid as important reaction parameters, and indane **2b** was generated in high yield under the optimal conditions (eq 2).[7]



With an effective racemic method in hand, we evaluated a variety of chiral ligands for their potential in the asymmetric arylcyanation of benzonitrile **1b** (Scheme 1). Whereas achiral monophosphine ligands provided high reaction rates and yields in the formation of product **2b**, reactions using representative chiral monophosphines were generally low-yielding and displayed poor enantioselectivity. Bidentate ligands such as (*R,R*)-Me-BPE,[8] (*R,R*)-BozPHOS,[9] (*R*)-*i*-Pr-PHOX,[10] and (*S,S,R,R*)-TangPHOS[11] proved more promising, affording **2b** in moderate to high enantioselectivities.

Highest ee's were achieved using TangPHOS as the chiral ligand, but product was generated in very low yield.[12] Examination of the crude reaction mixture revealed olefin isomerization of substrate **1b** as a major competing pathway.[13] Further, ^{1}H NMR studies of the reaction of Ni(cod)$_2$ and ligand to form Ni(cod)(TangPHOS) revealed that all released 1,5-cyclooctadiene had undergone isomerization to 1,3-cyclooctadiene. We reasoned that olefin isomerization

Scheme 1. Representative Chiral Ligands Screened in the Asymmetric Arylcyanation Reaction



of **1b** might be suppressed by using a different Ni^0 source. Indeed, use of NiCl$_2$·DME and Zn as the Ni^0 source[14] and increasing the ratio of TangPHOS to Ni led to minimized olefin isomerization and provided indane **2b** in 81% yield and 95% ee (eq 3).[15] Of the large number of boron- and aluminum-centered Lewis acids examined, BPh$_3$ afforded highest enantioselectivities.

Investigation of the scope of the arylcyanation reaction revealed that this methodology provides access to a range of substituted indane structures in highly enantioenriched form (Table 1). Substrates bearing varying substitution on the benzonitrile (R^1, entries 2 and 3) and on the alkene (R^2, entries 4−8) all underwent cyclization with consistently high ee's (92−96%). However, attainment of useful product yields from substrates bearing sterically demanding or electron-deficient alkene substituents necessitated elevated catalyst loadings (10 mol% NiCl$_2$·DME, 18 mol % TangPHOS, 20 mol % BPh$_3$, and 20 mol% Zn) and extended reaction times.

Fused pyrrole **2j** was generated in 97% ee from the corresponding homoallylic pyrrole-2-carbonitrile (entry 9), demonstrating the applicability of this method to heteroaromatic frameworks. While benzopyran **2k** could be accessed in 77% ee (entry 10), treatment of the analogous allylic ether under similar reaction conditions failed to provide the desired cyclization product **2l**. In fact, introduction of the same allylic ether to the arylcyanation of substrate **1b** led to complete catalyst inhibition and no detectable formation of indane **2b**. These observations are consistent with formation of an inactive π-allyl−nickel complex upon the addition of the electron-rich catalyst to allylic ethers.

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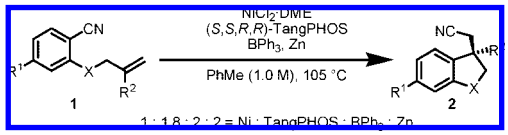
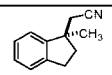
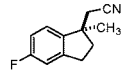
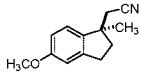
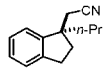
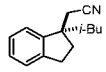
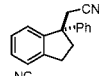
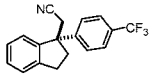
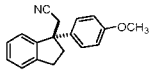
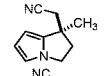
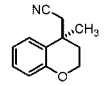
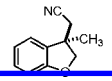
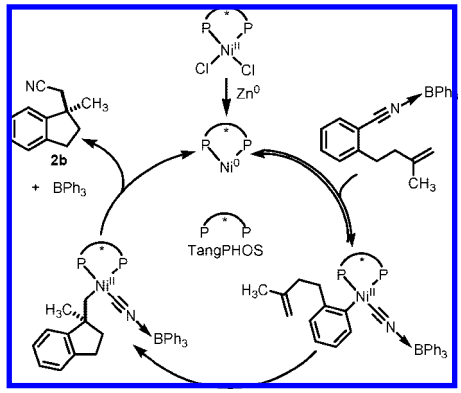
Table 1. Substrate Scope of the Asymmetric Olefin Arylcyanation Reaction[a]



Entry	Product		[Ni] (mol%)	Time (h)	Yield (%)[b]	Ee (%)[c]
1		**2b**	5	40	85	93
2		**2c**	5	40	84	92
3		**2d**	5	90	75	93
4		**2e**	10	90	69	94
5		**2f**	10	90	75	95
6		**2g**	10	90	72	95
7		**2h**	10	90	49	92
8		**2i**	10	40	65	96
9		**2j**	10	40	77	97
10[d]		**2k**	10	90	47	77
11[d]		**2l**	5	40	0	n.d.[e]

[a] Reactions carried out on 0.6 mmol scale unless noted otherwise. [b] Isolated yield after chromatography. [c] Determined by GC or HPLC analysis (see Supporting Information). [d] Reaction carried out on 0.3 mmol scale. [e] n.d. = Not determined.

Scheme 2. Proposed Catalytic Cycle



A likely mechanistic scenario for the catalytic arylcyanation is outlined in Scheme 2. The observed effects of substituents on the overall reaction rate are consistent with a mechanism involving Lewis acid coordination to the nitrile, with activation toward oxidative addition across the $C_{aryl}-CN$ bond by the Ni(0) complex.[16] Subsequent migratory insertion leads to generation of the $C_{aryl}-C_{quat}$ bond, and reductive elimination then results in formation of the $C_{sp3}-CN$ bond and regeneration of the Ni(0) catalyst. Because of the significant effect of olefin substituents on the reaction rate, it seems unlikely that oxidative addition is rate-determining, but the possibility that olefin−Ni coordination occurs prior to rate-determining oxidative addition cannot be excluded. The likelihood that BPh_3 remains coordinated to the CN fragment through the enantioselectivity-determining step is suggested by the strong dependence of enantioselectivity on the identity of the Lewis acid co-catalyst.

In summary, highly enantioselective, intramolecular alkene arylcyanation via C−CN bond activation has been accomplished using a Ni(0) catalyst and BPh_3 co-catalyst. TangPHOS was found to provide high enantioselectivity in this transformation. Current efforts directed toward more complete mechanistic studies of this reaction as well as extension of the substrate scope are ongoing in our laboratory.

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Supporting Information Available: Complete experimental procedures, detailed optimization studies, and characterization data for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

References

(1) Murakami, M.; Ito, Y. In *Activation of Unreactive Bonds and Organic Synthesis*; Murai, S., Ed.; Springer: Berlin, 1999; p 97.
(2) (a) Nakao, Y.; Hiyama, T. *Pure Appl. Chem.* **2008**, *80*, 1097. (b) Nakao, Y.; Yada, A.; Ebata, S.; Hiyama, T. *J. Am. Chem. Soc.* **2007**, *129*, 2428. (c) Nakao, Y.; Oda, S.; Yada, A.; Hiyama, T. *Tetrahedron* **2006**, *62*, 7567. (d) Nakao, Y.; Oda, S.; Hiyama, T. *J. Am. Chem. Soc* **2004**, *126*, 13904.
(3) Nakao, Y.; Yada, A.; Satoh, J.; Ebata, S.; Oda, S.; Hiyama, T. *Chem. Lett.* **2006**, *35*, 790.
(4) For examples of other Ni-catalyzed carbocyanation reactions, see: (a) Nakao, Y.; Hirata, Y.; Tanaka, M.; Hiyama, T. *Angew. Chem., Int. Ed.* **2008**, *47*, 385. (b) Nakao, Y.; Yukawa, T.; Hirata, Y.; Oda, S.; Satoh, J.; Hiyama, T. *J. Am. Chem. Soc.* **2006**, *128*, 7116.
(5) For recent reviews on quaternary carbon formation, see: (a) Douglas, C. J.; Overman, L. E. *Proc. Nat. Acad. Sci. U.S.A.* **2004**, *101*, 5363. (b) Trost, B. M.; Jiang, C. *Synthesis* **2006**, 369.
(6) For examples of palladium-catalyzed olefin and allene acylcyanation, including asymmetric variants, see: (a) Yasui, Y.; Kamisaki, H.; Takemoto, Y. *Org. Lett.* **2008**, *10*, 3303. (b) Kobayashi, Y.; Kamisaki, H.; Takeda, H.; Yasui, Y.; Yanada, R.; Takemoto, Y. *Tetrahedron* **2007**, *63*, 2978. (c) Kobayashi, Y.; Kamisaki, H.; Yanada, R.; Takemoto, Y. *Org. Lett.* **2006**, *8*, 2711. (d) Nishihara, Y.; Inoue, Y.; Izawa, S.; Miyasaka, M.; Tanemura, K.; Nakajima, K.; Takagi, K. *Tetrahedron* **2006**, *62*, 9872. (e) Nakao, Y.; Hirata, Y.; Hiyama, T. *J. Am. Chem. Soc.* **2006**, *128*, 7420.
(7) Lewis acid co-catalysts have previously been reported to accelerate alkyne arylcyanation reactions (see ref 2b).
(8) Burk, M. J.; Feaster, J. E.; Nugent, W. A.; Harlow, R. L. *J. Am. Chem. Soc.* **1993**, *115*, 10125.
(9) (a) Boezio, A. A.; Pytkowicz, J.; Côté, A.; Charette, A. B. *J. Am. Chem. Soc.* **2003**, *125*, 14260. (b) Côté, A.; Desrosiers, J.-N.; Boezio, A. A.; Charette, A. B. *Org. Synth.* **2006**, *83*, 1.
(10) Helmchen, G.; Pfaltz, A. *Acc. Chem. Res.* **2000**, *33*, 336.
(11) Tang, W.; Zhang, X. *Angew. Chem., Int. Ed.* **2002**, *41*, 1612.
(12) Other P-stereogenic ligands provided lower enantioselectivity. Details are included in the Supporting Information.
(13) In contrast, very little olefin isomerization was observed when *i*-Pr-PHOX was used. The mass balance was remaining starting material. Efforts to increase the yield of the reaction catalyzed by PHOX ligands were unsuccessful. In addition, use of $NiCl_2$·DME and Zn with *i*-Pr-PHOX led to only 22% yield of **2b** in 27% ee.
(14) (a) Percec, V.; Bae, J.-Y.; Zhao, M.; Hill, D. H. *J. Org. Chem.* **1995**, *60*, 176. (b) Percec, V.; Bae, J.-Y.; Hill, D. H. *J. Org. Chem.* **1995**, *60*, 1060.
(15) Excess TangPHOS may be required due to the presence of $ZnCl_2$, which forms upon reduction of $NiCl_2$·DME and may bind TangPHOS.
(16) (a) Huang, J.; Haar, C. M.; Nolan, S. P.; Marcone, J. E.; Moloy, K. G. *Organometallics* **1999**, *18*, 297. (b) Garcia, J. J.; Brunkan, N. M.; Jones, W. D. *J. Am. Chem. Soc.* **2002**, *124*, 9547. (c) Ateşin, T. A.; Lachaize, S.; García, J. J.; Jones, W. D. *Organometallics* **2008**, *27*, 3811.

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